organic compounds

3-(Phenylimino)indolin-2-one

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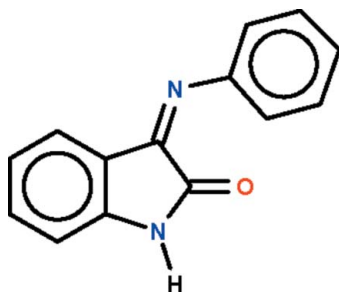Correspondence e-mail: seikweng@um.edu.my

The imino C=N double bond in the title compound, $C_{14}H_{10}N_2O$, exists in an E conformation, with the phenyl ring being twisted by 80.7 (1)° in one independent molecule and by 81.4 (1)° in the other with respect to the plane of the indoline fused-ring system. The two independent molecules are linked by N—H⋯O hydrogen bonds, forming a zigzag chain running along the a axis.

Related literature

For the synthesis, see: Grimshaw & Begley (1974). For the crystal structures of other phenyl-substituted derivatives, see: Akkurt *et al.* (2003); Hökelek *et al.* (2006); Öztürk *et al.* (2003).

Experimental

Crystal data

$C_{14}H_{10}N_2O$	V = 2215.77 (8) Å^3
M_r = 222.24	Z = 8
Orthorhombic, $Pna2_1$	Mo $K\alpha$ radiation
a = 20.1647 (4) Å	μ = 0.09 mm^{-1}
b = 5.0223 (1) Å	T = 293 K
c = 21.8791 (5) Å	$0.3 \times 0.3 \times 0.3$ mm

Data collection

Bruker APEXII diffractometer	2042 reflections with $I > 2\sigma(I)$
15055 measured reflections	R_{int} = 0.030
2619 independent reflections	

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.034	H atoms treated by a mixture of
$wR(F^2)$ = 0.118	independent and constrained
S = 1.10	refinement
2619 reflections	$\Delta\rho_{max}$ = 0.19 e Å^{-3}
315 parameters	$\Delta\rho_{min}$ = −0.20 e Å^{-3}
3 restraints	

Table 1
Hydrogen-bond geometry (Å, °).

D—H⋯A	D—H	H⋯A	D⋯A	D—H⋯A
N1—H1⋯N4[i]	0.86 (3)	2.11 (2)	2.940 (3)	161 (3)
N3—H3⋯N2	0.86 (3)	2.28 (3)	3.111 (3)	164 (3)

Symmetry code: (i) $x + \frac{1}{2}, -y + \frac{3}{2}, z$.

Data collection: *APEX2* (Bruker, 2005); cell refinement: *SAINT* (Bruker, 2005); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2010).

References

Akkurt, M., Öztürk, S., Erçağ, A., Özgür, M. Ü. & Heinemann, F. W. (2003). *Acta Cryst.* E**59**, o780–o782.
Barbour, L. J. (2001). *J. Supramol. Chem.* **1**, 189–191.
Bruker (2005). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
Grimshaw, J. & Begley, W. J. (1974). *Synthesis*, pp. 496–498.
Hökelek, T., Erçag, A., Çoruh, U., Vazquez-Lopez, E. M. & Özgür, M. Ü. (2006). *Anal. Sci. X-Ray Struct. Anal. Online*, **21**, x129–x130.
Öztürk, S., Akkurt, M., Özgür, M. Ü., Erçağ, A. & Heinemann, F. W. (2003). *Acta Cryst.* E**59**, o569–o571.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.
Westrip, S. P. (2010). *publCIF*. In preparation.

supplementary materials

3-(Phenylimino)indolin-2-one

A. Al Subari, R. Bouhfid, H. Zouihri, E. M. Essassi and S. W. Ng

Experimental

The compound was synthesized from the reaction of isatin and aniline in ethanol according to a literature method (Grimshaw & Begley, 1974), and crystals were obtained by recrystallization from ethanol.

Refinement

Carbon-bound H-atoms were placed in calculated positions (C—H 0.93 Å) and were included in the refinement in the riding model approximation, with U(H) set to $1.2U$(C). The amino H-atoms were located in a difference Fourier map, and were refined isotropically with a distance restraint of N–H 0.86±0.01 Å. Friedel pairs were merged.

Figures
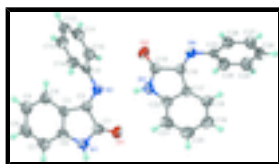
Fig. 1. Anisotropic displacement ellipsoid plot (Barbour, 2001) of the two independent molecules of $C_{14}H_{10}N_2O$ at the 50% probability level; hydrogen atoms are drawn as spheres of an arbitrary radius.

3-(Phenylimino)indolin-2-one

Crystal data

$C_{14}H_{10}N_2O$	$F(000) = 928$
$M_r = 222.24$	$D_x = 1.332$ Mg m^{-3}
Orthorhombic, $Pna2_1$	Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Hall symbol: P 2c -2n	Cell parameters from 3521 reflections
$a = 20.1647$ (4) Å	$\theta = 2.7–25.7°$
$b = 5.0223$ (1) Å	$\mu = 0.09$ mm^{-1}
$c = 21.8791$ (5) Å	$T = 293$ K
$V = 2215.77$ (8) Å^3	Block, orange
$Z = 8$	$0.3 \times 0.3 \times 0.3$ mm

Data collection

Bruker APEXII diffractometer	2042 reflections with $I > 2\sigma(I)$
Radiation source: fine-focus sealed tube	$R_{int} = 0.030$
graphite	$\theta_{max} = 27.5°$, $\theta_{min} = 1.9°$
φ and ω scans	$h = -24 \rightarrow 26$

15055 measured reflections

2619 independent reflections

$k = -3 \rightarrow 6$

$l = -28 \rightarrow 28$

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.034$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.118$	H atoms treated by a mixture of independent and constrained refinement
$S = 1.10$	$w = 1/[\sigma^2(F_o^2) + (0.0745P)^2 + 0.0171P]$ where $P = (F_o^2 + 2F_c^2)/3$
2619 reflections	$(\Delta/\sigma)_{max} = 0.001$
315 parameters	$\Delta\rho_{max} = 0.19$ e Å^{-3}
3 restraints	$\Delta\rho_{min} = -0.20$ e Å^{-3}

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	y	z	U_{iso}*/U_{eq}
O1	0.36628 (12)	0.9072 (6)	0.49888 (10)	0.0769 (7)
O2	0.11993 (12)	0.6636 (6)	0.52675 (12)	0.0816 (8)
N1	0.44843 (11)	0.6264 (6)	0.53285 (11)	0.0543 (6)
N2	0.28865 (11)	0.6607 (5)	0.59271 (10)	0.0505 (6)
N3	0.20427 (11)	0.9352 (5)	0.49173 (10)	0.0535 (6)
N4	0.04413 (11)	0.9133 (5)	0.43186 (11)	0.0525 (6)
C1	0.38742 (14)	0.7371 (6)	0.53269 (13)	0.0535 (7)
C2	0.34907 (12)	0.5996 (5)	0.58405 (11)	0.0462 (6)
C3	0.39537 (12)	0.4079 (5)	0.61115 (11)	0.0446 (6)
C4	0.39058 (14)	0.2185 (6)	0.65685 (13)	0.0519 (7)
H4	0.3517	0.2014	0.6794	0.062*
C5	0.44415 (16)	0.0553 (6)	0.66857 (15)	0.0605 (7)
H5	0.4414	−0.0724	0.6993	0.073*
C6	0.50158 (16)	0.0802 (7)	0.63515 (17)	0.0663 (8)
H6	0.5370	−0.0327	0.6436	0.080*
C7	0.50835 (15)	0.2678 (7)	0.58932 (15)	0.0609 (8)
H7	0.5475	0.2840	0.5672	0.073*
C8	0.45450 (13)	0.4302 (6)	0.57769 (12)	0.0481 (6)
C9	0.25376 (12)	0.5309 (5)	0.64115 (12)	0.0454 (6)
C10	0.20928 (13)	0.3331 (7)	0.62764 (14)	0.0565 (7)
H10	0.2017	0.2850	0.5872	0.068*
C11	0.17563 (15)	0.2051 (7)	0.67433 (17)	0.0619 (8)
H11	0.1464	0.0673	0.6653	0.074*
C12	0.18528 (14)	0.2810 (7)	0.73385 (15)	0.0594 (8)
H12	0.1628	0.1938	0.7651	0.071*
C13	0.22774 (15)	0.4843 (6)	0.74728 (13)	0.0565 (7)
H13	0.2334	0.5381	0.7876	0.068*

C14	0.26251 (16)	0.6108 (6)	0.70102 (13)	0.0536 (7)
H14	0.2916	0.7488	0.7102	0.064*
C15	0.14226 (13)	0.8292 (7)	0.49218 (13)	0.0539 (7)
C16	0.10494 (12)	0.9666 (5)	0.44047 (11)	0.0455 (6)
C17	0.15210 (13)	1.1524 (5)	0.41286 (11)	0.0437 (6)
C18	0.14711 (14)	1.3414 (6)	0.36677 (13)	0.0528 (7)
H18	0.1080	1.3596	0.3447	0.063*
C19	0.20099 (16)	1.5019 (6)	0.35414 (15)	0.0609 (7)
H19	0.1983	1.6295	0.3235	0.073*
C20	0.25942 (17)	1.4720 (7)	0.38751 (17)	0.0661 (8)
H20	0.2955	1.5803	0.3785	0.079*
C21	0.26500 (16)	1.2858 (7)	0.43355 (16)	0.0620 (8)
H21	0.3042	1.2679	0.4555	0.074*
C22	0.21093 (13)	1.1269 (5)	0.44624 (12)	0.0472 (6)
C23	0.00927 (12)	1.0342 (6)	0.38269 (12)	0.0461 (6)
C24	0.01763 (15)	0.9423 (6)	0.32334 (14)	0.0576 (7)
H24	0.0478	0.8065	0.3154	0.069*
C25	−0.01848 (16)	1.0510 (7)	0.27619 (16)	0.0610 (8)
H25	−0.0126	0.9887	0.2365	0.073*
C26	−0.06293 (15)	1.2504 (7)	0.28746 (16)	0.0630 (8)
H26	−0.0869	1.3255	0.2555	0.076*
C27	−0.07213 (15)	1.3401 (7)	0.34652 (17)	0.0687 (9)
H27	−0.1025	1.4755	0.3541	0.082*
C28	−0.03689 (14)	1.2315 (7)	0.39430 (14)	0.0586 (7)
H28	−0.0441	1.2903	0.4341	0.070*
H1	0.4791 (10)	0.653 (6)	0.5061 (10)	0.052 (8)*
H3	0.2338 (12)	0.883 (7)	0.5172 (12)	0.064 (9)*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0687 (14)	0.0941 (18)	0.0678 (15)	0.0084 (12)	0.0068 (11)	0.0340 (13)
O2	0.0684 (14)	0.104 (2)	0.0721 (14)	−0.0040 (13)	−0.0001 (12)	0.0422 (15)
N1	0.0457 (13)	0.0711 (16)	0.0461 (12)	−0.0002 (11)	0.0104 (10)	0.0017 (12)
N2	0.0423 (12)	0.0625 (15)	0.0468 (12)	0.0047 (10)	0.0007 (9)	0.0070 (11)
N3	0.0483 (12)	0.0653 (15)	0.0470 (12)	0.0081 (11)	−0.0064 (10)	0.0028 (11)
N4	0.0419 (12)	0.0659 (15)	0.0498 (12)	0.0040 (11)	0.0037 (9)	0.0097 (11)
C1	0.0477 (15)	0.0673 (19)	0.0454 (15)	−0.0002 (13)	0.0020 (12)	0.0021 (14)
C2	0.0440 (13)	0.0533 (15)	0.0414 (12)	−0.0015 (11)	−0.0017 (10)	−0.0023 (12)
C3	0.0424 (13)	0.0466 (14)	0.0447 (13)	0.0000 (11)	−0.0018 (10)	−0.0070 (11)
C4	0.0528 (15)	0.0507 (15)	0.0523 (16)	−0.0015 (12)	−0.0007 (12)	−0.0013 (13)
C5	0.0673 (18)	0.0514 (16)	0.0628 (17)	0.0045 (14)	−0.0089 (15)	0.0015 (14)
C6	0.0586 (17)	0.0610 (18)	0.079 (2)	0.0137 (15)	−0.0129 (15)	−0.0068 (17)
C7	0.0498 (16)	0.0683 (19)	0.0645 (19)	0.0081 (13)	0.0051 (14)	−0.0094 (16)
C8	0.0452 (14)	0.0521 (15)	0.0471 (13)	0.0008 (11)	0.0006 (11)	−0.0098 (12)
C9	0.0355 (12)	0.0527 (15)	0.0479 (13)	0.0082 (11)	0.0022 (10)	0.0033 (12)
C10	0.0534 (16)	0.0620 (18)	0.0541 (16)	−0.0013 (14)	−0.0060 (14)	−0.0022 (14)
C11	0.0501 (16)	0.0592 (17)	0.076 (2)	−0.0087 (14)	−0.0041 (15)	0.0083 (16)

C12	0.0505 (17)	0.0612 (18)	0.067 (2)	0.0046 (14)	0.0100 (14)	0.0159 (15)
C13	0.0664 (17)	0.0566 (16)	0.0465 (14)	0.0090 (14)	0.0075 (13)	−0.0021 (13)
C14	0.0539 (16)	0.0525 (16)	0.0544 (16)	−0.0026 (13)	0.0024 (12)	−0.0016 (12)
C15	0.0486 (15)	0.0687 (19)	0.0444 (15)	0.0099 (14)	0.0045 (12)	0.0066 (13)
C16	0.0411 (13)	0.0532 (15)	0.0422 (13)	0.0076 (11)	0.0045 (10)	0.0025 (11)
C17	0.0444 (14)	0.0452 (14)	0.0414 (13)	0.0083 (11)	0.0055 (10)	−0.0025 (10)
C18	0.0525 (16)	0.0513 (15)	0.0545 (17)	0.0073 (12)	0.0051 (13)	0.0036 (13)
C19	0.0660 (18)	0.0527 (16)	0.0640 (17)	0.0002 (14)	0.0109 (15)	0.0083 (15)
C20	0.0609 (17)	0.0572 (18)	0.080 (2)	−0.0122 (15)	0.0092 (17)	−0.0001 (17)
C21	0.0492 (16)	0.0621 (18)	0.075 (2)	−0.0016 (13)	−0.0026 (15)	−0.0090 (17)
C22	0.0468 (14)	0.0480 (14)	0.0467 (13)	0.0062 (11)	0.0026 (11)	−0.0062 (12)
C23	0.0355 (12)	0.0510 (15)	0.0519 (14)	0.0011 (11)	0.0023 (10)	0.0069 (12)
C24	0.0606 (17)	0.0549 (17)	0.0572 (16)	0.0146 (13)	0.0025 (14)	0.0000 (14)
C25	0.0688 (19)	0.0606 (18)	0.0536 (15)	0.0011 (15)	−0.0094 (14)	−0.0036 (15)
C26	0.0542 (18)	0.0700 (19)	0.065 (2)	0.0061 (15)	−0.0105 (14)	0.0155 (16)
C27	0.0550 (18)	0.069 (2)	0.082 (2)	0.0236 (15)	0.0052 (16)	0.0144 (18)
C28	0.0542 (16)	0.0647 (19)	0.0569 (17)	0.0108 (14)	0.0077 (14)	0.0011 (14)

Geometric parameters (Å, °)

O1—C1	1.208 (4)	C11—H11	0.9300
O2—C15	1.211 (4)	C12—C13	1.364 (5)
N1—C1	1.350 (4)	C12—H12	0.9300
N1—C8	1.396 (4)	C13—C14	1.386 (4)
N1—H1	0.86 (1)	C13—H13	0.9300
N2—C2	1.271 (3)	C14—H14	0.9300
N2—C9	1.429 (3)	C15—C16	1.524 (4)
N3—C15	1.359 (4)	C16—C17	1.463 (4)
N3—C22	1.391 (4)	C17—C18	1.389 (4)
N3—H3	0.86 (1)	C17—C22	1.399 (4)
N4—C16	1.269 (3)	C18—C19	1.381 (4)
N4—C23	1.421 (3)	C18—H18	0.9300
C1—C2	1.529 (4)	C19—C20	1.394 (5)
C2—C3	1.466 (4)	C19—H19	0.9300
C3—C4	1.383 (4)	C20—C21	1.379 (5)
C3—C8	1.404 (4)	C20—H20	0.9300
C4—C5	1.380 (4)	C21—C22	1.380 (4)
C4—H4	0.9300	C21—H21	0.9300
C5—C6	1.375 (5)	C23—C28	1.383 (4)
C5—H5	0.9300	C23—C24	1.389 (4)
C6—C7	1.383 (5)	C24—C25	1.376 (4)
C6—H6	0.9300	C24—H24	0.9300
C7—C8	1.382 (4)	C25—C26	1.366 (4)
C7—H7	0.9300	C25—H25	0.9300
C9—C10	1.371 (4)	C26—C27	1.381 (5)
C9—C14	1.381 (4)	C26—H26	0.9300
C10—C11	1.385 (4)	C27—C28	1.377 (4)
C10—H10	0.9300	C27—H27	0.9300
C11—C12	1.371 (5)	C28—H28	0.9300

C1—N1—C8	111.9 (2)	C14—C13—H13	119.9
C1—N1—H1	126 (2)	C9—C14—C13	119.7 (3)
C8—N1—H1	121 (2)	C9—C14—H14	120.2
C2—N2—C9	118.2 (2)	C13—C14—H14	120.2
C15—N3—C22	111.4 (2)	O2—C15—N3	128.0 (3)
C15—N3—H3	121 (2)	O2—C15—C16	126.2 (3)
C22—N3—H3	128 (2)	N3—C15—C16	105.7 (2)
C16—N4—C23	120.0 (2)	N4—C16—C17	134.5 (2)
O1—C1—N1	127.9 (3)	N4—C16—C15	119.4 (2)
O1—C1—C2	126.2 (3)	C17—C16—C15	105.9 (2)
N1—C1—C2	105.8 (2)	C18—C17—C22	120.2 (3)
N2—C2—C3	135.1 (3)	C18—C17—C16	133.5 (3)
N2—C2—C1	119.0 (2)	C22—C17—C16	106.1 (2)
C3—C2—C1	105.8 (2)	C19—C18—C17	119.2 (3)
C4—C3—C8	119.4 (3)	C19—C18—H18	120.4
C4—C3—C2	134.4 (2)	C17—C18—H18	120.4
C8—C3—C2	106.1 (2)	C18—C19—C20	119.8 (3)
C5—C4—C3	119.2 (3)	C18—C19—H19	120.1
C5—C4—H4	120.4	C20—C19—H19	120.1
C3—C4—H4	120.4	C21—C20—C19	121.6 (3)
C6—C5—C4	120.4 (3)	C21—C20—H20	119.2
C6—C5—H5	119.8	C19—C20—H20	119.2
C4—C5—H5	119.8	C20—C21—C22	118.3 (3)
C5—C6—C7	122.0 (3)	C20—C21—H21	120.8
C5—C6—H6	119.0	C22—C21—H21	120.8
C7—C6—H6	119.0	C21—C22—N3	128.4 (3)
C8—C7—C6	117.3 (3)	C21—C22—C17	120.8 (3)
C8—C7—H7	121.4	N3—C22—C17	110.8 (2)
C6—C7—H7	121.4	C28—C23—C24	119.5 (3)
C7—C8—N1	128.0 (3)	C28—C23—N4	120.0 (3)
C7—C8—C3	121.6 (3)	C24—C23—N4	120.4 (3)
N1—C8—C3	110.4 (2)	C25—C24—C23	120.3 (3)
C10—C9—C14	119.9 (3)	C25—C24—H24	119.8
C10—C9—N2	119.5 (3)	C23—C24—H24	119.8
C14—C9—N2	120.5 (3)	C26—C25—C24	120.2 (3)
C9—C10—C11	119.9 (3)	C26—C25—H25	119.9
C9—C10—H10	120.1	C24—C25—H25	119.9
C11—C10—H10	120.1	C25—C26—C27	119.7 (3)
C12—C11—C10	120.1 (3)	C25—C26—H26	120.1
C12—C11—H11	119.9	C27—C26—H26	120.1
C10—C11—H11	119.9	C28—C27—C26	120.8 (3)
C13—C12—C11	120.1 (3)	C28—C27—H27	119.6
C13—C12—H12	119.9	C26—C27—H27	119.6
C11—C12—H12	119.9	C27—C28—C23	119.5 (3)
C12—C13—C14	120.2 (3)	C27—C28—H28	120.3
C12—C13—H13	119.9	C23—C28—H28	120.3
C8—N1—C1—O1	−178.4 (3)	C22—N3—C15—O2	−178.1 (3)
C8—N1—C1—C2	0.7 (3)	C22—N3—C15—C16	0.7 (3)
C9—N2—C2—C3	3.4 (5)	C23—N4—C16—C17	5.7 (5)

C9—N2—C2—C1	−179.7 (2)	C23—N4—C16—C15	−178.2 (2)
O1—C1—C2—N2	1.5 (5)	O2—C15—C16—N4	1.7 (5)
N1—C1—C2—N2	−177.6 (3)	N3—C15—C16—N4	−177.1 (3)
O1—C1—C2—C3	179.3 (3)	O2—C15—C16—C17	178.9 (3)
N1—C1—C2—C3	0.1 (3)	N3—C15—C16—C17	0.1 (3)
N2—C2—C3—C4	0.2 (6)	N4—C16—C17—C18	0.4 (5)
C1—C2—C3—C4	−177.1 (3)	C15—C16—C17—C18	−176.1 (3)
N2—C2—C3—C8	176.3 (3)	N4—C16—C17—C22	175.8 (3)
C1—C2—C3—C8	−0.9 (3)	C15—C16—C17—C22	−0.8 (3)
C8—C3—C4—C5	0.0 (4)	C22—C17—C18—C19	0.4 (4)
C2—C3—C4—C5	175.7 (3)	C16—C17—C18—C19	175.2 (3)
C3—C4—C5—C6	−0.1 (4)	C17—C18—C19—C20	0.1 (4)
C4—C5—C6—C7	0.4 (5)	C18—C19—C20—C21	−0.4 (5)
C5—C6—C7—C8	−0.6 (5)	C19—C20—C21—C22	0.0 (5)
C6—C7—C8—N1	−177.6 (3)	C20—C21—C22—N3	−177.2 (3)
C6—C7—C8—C3	0.4 (4)	C20—C21—C22—C17	0.6 (4)
C1—N1—C8—C7	176.8 (3)	C15—N3—C22—C21	176.7 (3)
C1—N1—C8—C3	−1.4 (3)	C15—N3—C22—C17	−1.2 (3)
C4—C3—C8—C7	−0.1 (4)	C18—C17—C22—C21	−0.8 (4)
C2—C3—C8—C7	−177.0 (3)	C16—C17—C22—C21	−176.9 (2)
C4—C3—C8—N1	178.2 (2)	C18—C17—C22—N3	177.3 (2)
C2—C3—C8—N1	1.4 (3)	C16—C17—C22—N3	1.2 (3)
C2—N2—C9—C10	−104.2 (3)	C16—N4—C23—C28	−107.2 (3)
C2—N2—C9—C14	78.5 (4)	C16—N4—C23—C24	77.5 (4)
C14—C9—C10—C11	−3.1 (4)	C28—C23—C24—C25	1.6 (5)
N2—C9—C10—C11	179.5 (2)	N4—C23—C24—C25	176.9 (3)
C9—C10—C11—C12	1.9 (4)	C23—C24—C25—C26	−0.1 (5)
C10—C11—C12—C13	0.5 (5)	C24—C25—C26—C27	−0.9 (5)
C11—C12—C13—C14	−1.5 (5)	C25—C26—C27—C28	0.2 (5)
C10—C9—C14—C13	2.1 (4)	C26—C27—C28—C23	1.4 (5)
N2—C9—C14—C13	179.4 (2)	C24—C23—C28—C27	−2.3 (5)
C12—C13—C14—C9	0.3 (4)	N4—C23—C28—C27	−177.6 (3)

Hydrogen-bond geometry (Å, °)

D—H⋯A	D—H	H⋯A	D⋯A	D—H⋯A
N1—H1⋯N4[i]	0.86 (3)	2.11 (2)	2.940 (3)	161 (3)
N3—H3⋯N2	0.86 (3)	2.28 (3)	3.111 (3)	164 (3)

Symmetry codes: (i) $x+1/2$, $−y+3/2$, z.

Fig. 1